

## SPECTROPHOTOMETRIC DETERMINATION OF DRUGS AND PHARMACEUTICALS BY USING FOLIN-CIOCALTEU REAGENT

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### ABSTARCT

A simple, fast, selective and accurate spectrophotometric method has been outlined for the assessment of five phenolic drugs namely Balsalazide, Desvenlafaxine, Formoterol, Nelfinavir mesylate, Xipamide. This method is based on the development of blue coloured chromogen due to reduction of tungstate and/ or molybdate in Folin-Ciocalteu reagent by Drug in alkaline medium. The coloured species has an absorption maximum at 770nm, 755nm, 735nm, 760nm, 740nm and the Beer's law obeys over the concentration range of 7–48µg/mL, 2-14 µg/mL, 5-35 µg/mL, 3.5-24.5 µg/mL, 11-77 µg/mL respectively for the drugs Balsalazide, Desvenlafaxine, Formoterol, Nelfinavir mesylate, Xipamide. This method has been validated according to ICH guidelines and applied to the analysis of pharmaceuticals.

**KEYWORDS:** Spectrophotometry, Folin-Ciocalteu reagent, drugs, pharmaceutical formulations.

### 1. INTRODUCTION

1. Balsalazide [BSZ] is chemically known as (E)-5-[[4-[[[(2 carboxyethyl)amino]carboxyl]phenyl]azo]-2-hydroxy benzoic acid, disodium dihydrate. It is a prodrug that is enzymatically divided in the colon to deliver mesalamine (5-aminosalicylic acid), an anti-inflammatory drug (NSAID) which is useful for treatment of active ulcerative colons and gastrointestinal disorders such as inflammatory bowel disease. Balsalazide and one of its primary metabolites prevent the growth of cultivated human colon cancer cells and Balsalazide inhibit abnormal crypt formation in carcinogenic azoxymethane-treated animals. It is utilized in the treatment of mild to moderate ulcerative colitis.<sup>[1-3]</sup> Balsalazide disodium capsules comprise Balsalazide

disodium granule that are insoluble in acid and are meant to be administered to the colon intact. Once bacterial azo reductases enter the colon, break the compound to release 5-aminosalicylic acid, the molecule's therapeutically active component, and 4-aminobenzoyl- $\beta$ -alanine.<sup>[4]</sup> Literature survey found many analytical methods for the determination of Balsalazide. These methods include UV Spectrophotometric method<sup>[4-6]</sup>, HPLC<sup>[7-8]</sup> and Liquid Chromatography.<sup>[9]</sup>

2. Desvenlafaxine [DVL] is chemically designated as RS-4-[2-dimethylamino-1-(1 hydroxy cyclohexyl) ethyl]phenol. It is a serotonin-norepinephrine reuptake inhibitor class antidepressant, is used in the treatment of depression. Desvenlafaxine is a synthetic form of the isolated major active metabolite of venlafaxine and is grouped as a selective serotonin and norepinephrine reuptake inhibitor.<sup>[10]</sup> Literature survey reveals that methods have been reported for analysis of desvenlafaxine by UV-Visible spectroscopy<sup>[11-13]</sup>, HPTLC<sup>[14-15]</sup>, Liquid Chromatography<sup>[16]</sup>, UPLC-TMS<sup>[17]</sup>, LC-UV & LC-MS<sup>[18]</sup>, Spectrofluorimetric method<sup>[19]</sup>, Voltammetry.<sup>[20]</sup>

3. Formoterol [FMT] chemically known as N-[2-hydroxy-5-[1-hydroxy-2-[1-(4-methoxy phenyl) propan-2-ylamino]ethyl]phenyl]formamide. Formoterol is a long-acting  $\beta$ -2 agonist used for asthma and chronic pulmonary obstructive disease. It is a bronchodilator  $\beta$ -receptor with impressive Broncho selectivity, anti-allergic activity and pulmonary edema-inhibitory effect.<sup>[21-22]</sup> A survey of the literature revealed that a few analytical methods have been reported for the determination of formoterol by UV-Visible Spectrometry<sup>[22-23]</sup>, HPLC<sup>[24-27]</sup>, HPTLC<sup>[28]</sup>, Capillary Electrophoresis<sup>[29]</sup>, Gas Chromatography<sup>[30]</sup>, NMR spectroscopy.<sup>[31]</sup>

4. Nelfinavir mesylate (NFM) is a novel HIV-1 protease inhibitor. Its chemical name is (3S,4aS,8aS)-N-tert-butyl-2-[(2R,3R)-2-hydroxy-3-[(3-hydroxy-2methylphenyl) formamido]-4-(phenylsulfanyl)butyl]-decahydroisoquinoline-3-carboxamide methane sulfonate. It is the mesylate salt of a simple amine and has shown an effective and selective protease inhibitor for human immunodeficiency virus 1 (HIV-1). Nelfinavir mesylate has been shown to substantially reduce viral load and increase CD4+ T cells in adults and children infected with HIV, especially when administered in combination with other anti-HIV agents, nucleoside analogues, reverse transcriptase inhibitors and protease inhibitors.<sup>[32]</sup> NFM has been determined by many analytical methods, such as; Spectrophotometric method<sup>[33]</sup>, HPLC method<sup>[34-37]</sup>, HPTLC method<sup>[38]</sup>, LC-MS<sup>[39-40]</sup>, Liquid Chromatography<sup>[41]</sup>, RPLC.<sup>[42]</sup>

5. Xipamide (XPM) is chemically 4-chloro-N-(2,6-dimethylphenyl)-2-hydroxy-5-sulfamoylbenzamide. It is a non-thiazide diuretic with a more natriuretic effect than the thiazides and a less rapid onset and longer duration of action than furosemide. It has a moderately strong diuretic action. It lowers active sodium re-absorption and subsequent chloride by binding and inhibiting its action to the chloride site of the electro neutral Na<sup>+</sup>/Cl<sup>-</sup> co-transport system. Xipamide provides an effective alternative to other diuretics in the treatment of patients with mild to moderate hypertension and oedema patients for a variety of causes.<sup>[43]</sup> The literature survey reveals several analytical methods for the determination of Xipamide including Spectrophotometric.<sup>[44-47]</sup>, Spectrofluorimetric<sup>[46-48]</sup>, HPLC.<sup>[49-51]</sup>, HPLC-DAD<sup>[52]</sup>, Voltammetry.<sup>[53]</sup>

Chromatographic methods are the most widely used ones. Although various techniques listed above are specific, most of the described techniques are time consuming and require multistage extraction procedures. On the other hand, the mentioned spectrophotometric techniques are for numerous reasons not satisfactory for the routine quality assurance. Some of these techniques suffer from drawbacks such as poor sensitivity, low species stability, use of organic solvent, scrupulous monitoring of experimental variables and special equipment.

Folin-Ciocalteu (F-C) reagent is commonly used in plant biology to evaluate polyphenols and to assess other phenolic compounds in pharmaceuticals based on a reduction of Folin-Ciocalteu reagents. This paper outlines a reduction process involving F-C reagent and drugs leading to the formation of a blue coloured chromogen that can be tested at 740nm- 770nm range for respective drugs.

## STRUCTURE OF DRUGS

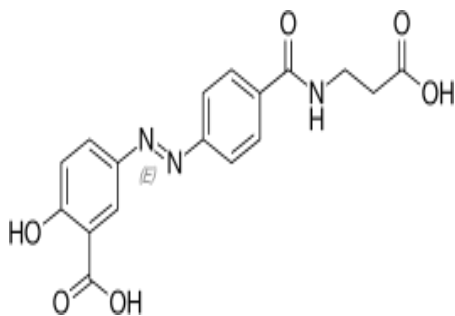


Fig 1: Balsalazide

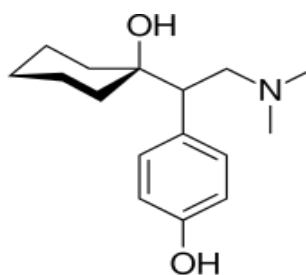
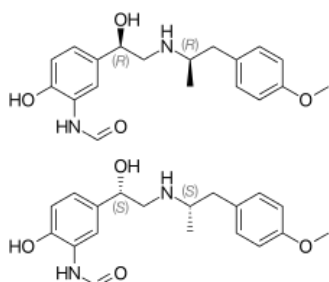
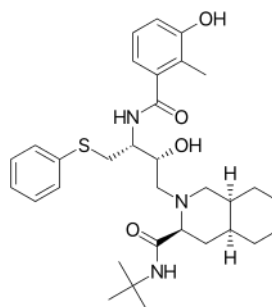
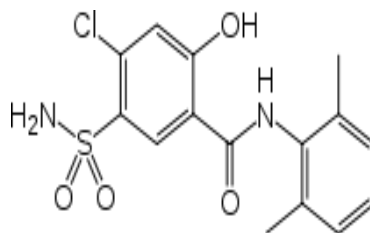


Fig 2: Desvenlafaxine

**Fig 3: Formoterol****Fig 4: Nelfinavir mesylate****Fig 5: Xipamide**

## 2. MATERIALS AND METHODS

### 2.1 Instrumentation

All absorption spectra were recorded on SHIMADZU 2600 UV-Visible double beam spectrophotometers using quartz cells of 10mm path length. A Dhona 200 single pan electrical balance is used for weighing the samples.

### 2.2 Materials and Reagents

All chemicals and reagents used were of analytical or pharmaceutical grade. Double distilled water was used throughout the investigation. Folin-Ciocalteu reagent (Merck, Mumbai), Sodium carbonate (Hetero Drugs Pvt Ltd, Hyd) were of analytical reagent grade and used without further purification. The pharmaceutical grade pure drug samples were kindly provided by GVK biosciences Pvt Ltd, Hetero Drugs Pvt Ltd, Dr Reddy's Laboratories, Hyderabad.

A stock standard solution of each drug containing 100 µg/mL were prepared by dissolving accurately weighed 10 mg of the respective drug transferred in to a 100 mL volumetric flask and made up to mark with distilled water. The solutions were further diluted quantitatively according to their linearity range. The pharmaceutical preparations were purchased from a local market and analysed.

### 2.3 Method development

Different aliquots of the drug solution (1 to 7 mL) were transferred into a series of 10 mL calibrated flasks. To each flask, 1 mL 10% Na<sub>2</sub>CO<sub>3</sub> was added, followed by 2 mL 2N Folin-Ciocalteu reagent. The contents were mixed and they were set aside for 20 minutes under occasional shaking. The volume was made up to the mark with water and the absorbance of each solution was measured at 740-770nm range against a reagent blank.

### 2.4 Construction of calibration curves

The calibration curve was plotted by taking concentration of the drugs in X-axis and absorbance in Y-axis. The calibration curves were constructed by taking absorbance data in six replicate experiments. The absorbance to concentration called relative response is calculated. Those points falling between 95% to 105% of the average relative response are only considered for construction of calibration.

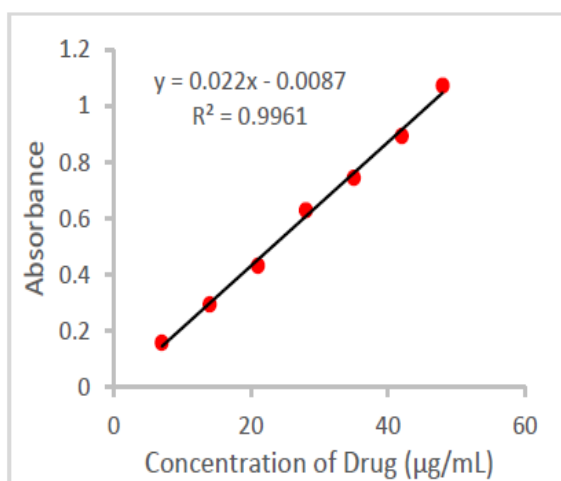


Fig 6: Calibration curve of Balsalazide

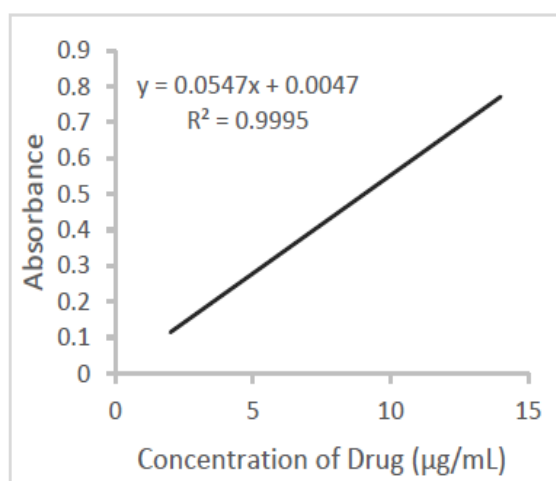


Fig 7: Calibration curve of Desvenlafaxine

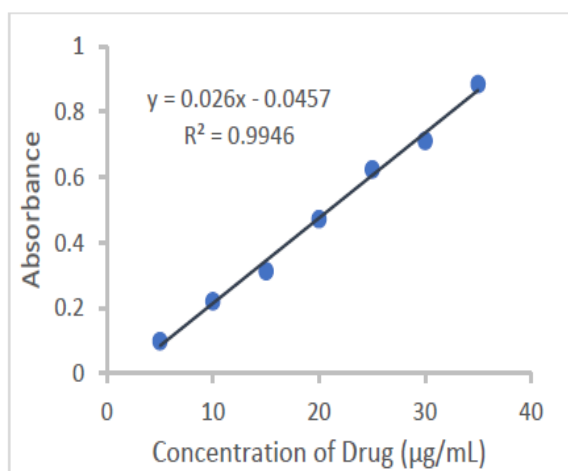


Fig 8: Calibration curve of Formoterol

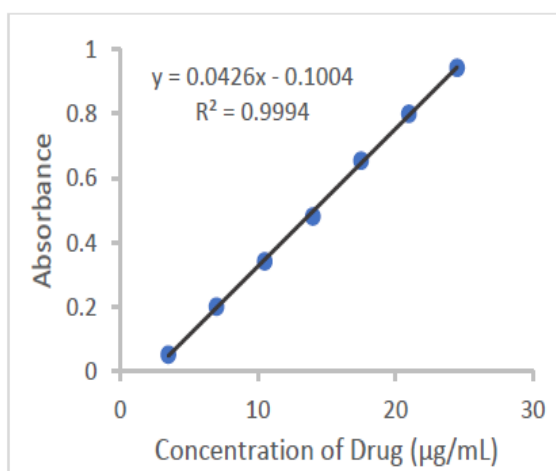
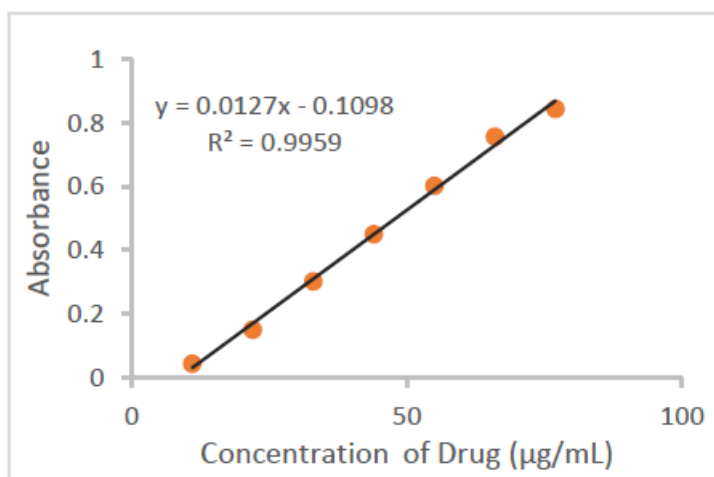


Fig 9: Calibration curve of Nelfinavir mesylate



**Fig 10: Calibration curve of Xipamide**

### 2.5 Accuracy and Precision studies

Accuracy of the methods developed are evaluated from the recovery studies on pure drug sample. At least four known concentration of drug solutions have been brought into Beer's law limit and recovery studies have been carried out. Excellent recovery showed the validity of the calibration curves for each drug. Precision of the procedure is illustrated by repeating experiment (n=6) and %RSD is found out. %RSD being less than 1 in each case speaks the high precision of the method.

### 2.6 Procedure for analysis of pharmaceuticals Balsalazide

Two tablets (BALACOL, 750mg) were weighed and crushed in to fine powder. Powder weight equivalent to 10mg of Balsalazide was transferred into 100mL calibrated flask and made up to mark with water. And the solution filtered through a Whatman No. 42 filter paper. It was used as stock sample solution and was further diluted with distilled water to get working concentration solution for analysis of the drug.

#### Desvenlafaxine

An amount of finely ground tablet powder (D-VENIZ, 50mg) equivalent to 10 mg of Desvenlafaxine was accurately weighed in 100 mL volumetric flask, the flask was shaken after addition of a 50 mL of water for about 5 minutes and the volume was made up to the mark with distilled water. The solution was kept aside for 20 minutes and filtered using Whatman No. 42 filter paper. The first 10 mL portion of the filtrate was thrown away and an appropriate aliquot as described under "General Analytical Procedure" was used for the assay.

**Formoterol**

For the determination of pharmaceutical formulations about 10 to 15 tablets (DERIFORM, 12mg) were weighed and powdered. A weight equivalent to 10mg of Formoterol was transferred into a 100 mL volumetric flask and finally volume was diluted up to the mark with distilled water, mixed well and the solution was kept aside for 20 minutes. After 20 minutes solution filtered using a Whatman No. 42 filter paper. First 10ml portion of the filtrate was discarded and a suitable aliquot of the subsequent portion was diluted appropriately to get working concentration and the assay was completed according to the procedure described above.

**Nelfinavir**

Three tablets (NEL, 250mg) were weighed and ground in to fine powder. A quantity equivalent to 10mg of Nelfinavir has been taken into a 100 mL of calibrated flask and added to about 30mL of methanol, sonicated for 30 minutes and filtered through Whatman filter paper No-42. The residue was washed thrice with methanol for complete recovery of drug and methanol was evaporated. The residue was dissolved in 100mL of distilled water. It was used as a stock sample solution. Aliquot portions of this stock solution were further diluted with distilled water to get the final concentration required for the determination of the drug.

**Xipamide**

About 15 to 20 tablets (XIPAMID, 20mg) were accurately weighed and crushed to a fine powder and the powder equivalent to 10mg of Xipamide was weighed accurately and transferred to 100mL volumetric flask, 60 mL of water was poured into the flask after that flask was shaken for about 5 minutes and the volume was made up to the mark with distilled water. The solution was kept aside for 20 minutes and filtered using Whatman No. 42 filter paper. First 10ml portion of the filtrate was discarded and an appropriate aliquot of the subsequent portion was diluted appropriately to get working concentration and the determination was performed according to the procedure described above.

The drug solutions extracted from tablet formulations were subjected to redox reaction with Folin-Ciocalteu reagent and subsequent determination was done. The concentration of tablet solutions falling under Beer's law limit was chosen for drug assay in the tablet. An outstanding tally between the concentration of drugs taken and found indicated the applicability of methods for formulations.

### 3. RESULTS AND DISCUSSIONS

The F-C reagent is used to evaluate several phenolic compounds and a great number of pharmaceutical-interest substances. The structural features of mentioned drugs permitted Folin-Ciocalteu reagent to be used for its analysis. The proposed technique is based on the development of a blue coloured chromogen, following the reduction of phosphor-molybd tungstic mixed acid of the F-C reagent by phenolic drugs, in the presence of sodium carbonate, which could be estimated at 740-770nm range for respective drugs. The mixed acids in the F-C reagent are the final chromogen and involve the following chemical species.

$3\text{H}_2\text{O} \cdot \text{P}_2\text{O}_5 \cdot 13 \text{WO}_3 \cdot 5\text{MoO}_3 \cdot 10\text{H}_2\text{O}$  and  $3\text{H}_2\text{O} \cdot \text{P}_2\text{O}_5 \cdot 14 \text{WO}_3 \cdot 4 \text{MoO}_3 \cdot 10\text{H}_2\text{O}$ .

The F-C reagent is assumed to contain heteropolyphospho-tungstatesmolybdates. Drugs are likely to reduce tungstate and/or molybdate oxygen atoms in the F-C reagent by producing one or more possible reduced species which have characteristic intense blue colour.

#### 3.1. Analytical data

Under optimum conditions a linear correlation was found between absorbance at  $\lambda_{\text{max}}$  and concentration of all drugs in the ranges given in table 1. Sensitivity parameters such as molar absorptivity, Sandell sensitivity are also presented in Table 1. Regression analysis of Beer's law data using the method of least squares was made to evaluate the slope (b), intercept (a), correlation coefficient (r) and is also given in table 1.

The LOD and LOQ were determined based on the standard deviation of the y-intercept and the slope of the calibration curves and presented in table1.

$\text{LOD} = 3.3 \text{ a} / \text{b}$   $\text{LOQ} = 10 \text{ a} / \text{b}$ .

Where a = standard deviation of the intercept (n = 6) b = slope of Calibration plot.

**Table 1: Analytical Parameters for determination of drugs By Red-Ox reaction with Folin-Ciocalteu reagent.**

Parameters	BSZ	DVL	FMT	NFN	XPM
$\lambda_{\text{max}}$ (nm)	770	755	735	760	740
Beer's law Limits ( $\mu\text{g/mL}$ )	7-48	2-14	5-35	3.5-24.5	11-77
Sandell sensitivity ( $\mu\text{g cm}^{-2}$ )	0.0454	0.018	0.038	0.0234	0.0787
Limit of Detection ( $\mu\text{g mL}^{-1}$ )	0.375	0.855	0.495	3.81	9.22
Limit of Quantification ( $\mu\text{g mL}^{-1}$ )	1.136	2.59	1.5	11.57	27.95



<b>Regression equation, Y</b>	=0.022x- 0.0087	=0.0547x+ 0.0047	=0.026x- 0.0457	=0.0426x- 0.1004	=0.0127x- 0.1098
<b>Intercept (a)</b>	0.0087	0.0047	0.0457	0.1004	0.1098
<b>Slope (b)</b>	0.022	0.0547	0.026	0.0426	0.0127
<b>Correlation coefficient (r)</b>	0.998	0.999	0.997	0.999	0.997
<b>SD of Intercept (Sa)</b>	0.0025	0.014	0.0039	0.0493	0.0355
<b>SD of Slope (Sb)</b>	0.0047	0.0064	0.0045	0.0016	0.034
<b>Molar Absorptivity (L mol<sup>-1</sup> cm<sup>-1</sup>)</b>	8.16×10 <sup>3</sup>	1.55×10 <sup>4</sup>	6.87×10 <sup>3</sup>	8.65×10 <sup>3</sup>	1.3×10 <sup>3</sup>

Table 2: Determination of accuracy and precision of the methods on pure drug samples.

Drug	Taken (µg/mL)	Found (µg/mL)	Error (%)	Recovery (%)	RSD (%)	Proposed method mean±SD
BSZ	14.0	14.02	0.14	100.14	0.154	100.06±0.155
	21.0	20.97	0.14	99.86		
	28.0	28.01	0.03	100.03		
	35.0	35.08	0.22	100.22		
DVL	6.0	6.02	0.33	100.33	0.48	100.08±0.481
	8.0	7.96	0.5	99.5		
	10.0	10.06	0.6	100.6		
	12.0	11.99	0.08	99.92		
FMT	10.0	9.97	0.3	99.7	0.261	99.98±0.261
	15.0	15.05	0.33	100.33		
	20.0	20.0	0.00	100.00		
	25.0	24.98	0.08	99.92		
NFN	7.0	6.99	0.14	99.86	0.222	99.85±0.222
	10.5	10.48	0.19	99.81		
	14.0	14.02	0.14	100.14		
	17.5	17.43	0.4	99.60		
XPM	22.0	21.98	0.09	99.91	0.081	100.01±0.082
	33.0	33.01	0.03	100.03		
	44.0	44.01	0.02	100.02		
	55.0	55.06	0.11	100.11		

Table 3: Results of assay of tablets by proposed method and statistical evaluation and recovery experiments by standard addition method.

Drug	Taken (µg/mL)	Found (µg/mL)	Error (%)	Recovery (%)	RSD (%)	Proposed method mean± SD	Reference method mean±SD	t- Test	F- Test
BSZ	14.0	13.98	0.14	99.86	0.122	99.95±0.122	99.09±0.106	0.25	1.32
	24.0	23.96	0.16	99.84					
	35.0	35.03	0.08	100.08					
	48.0	48.02	0.04	100.04					
DVL	4.0	4.00	0.0	100.00	0.212	99.97±0.212	99.26±0.277	0.413	0.592
	8.0	7.98	0.25	99.75					
	12.0	12.03	0.25	100.25					

	16.0	15.98	0.12	99.88					
<b>FMT</b>	12.0	12.00	0.00	100.00	0.136	100.02±0.137	100.62±0.587	1.31	0.052
	18.0	18.04	0.22	100.22					
	24.0	23.98	0.08	99.92					
	30.0	29.98	0.06	99.94					
<b>NFN</b>	5.0	5.00	0.0	100.00	0.035	99.97±0.0355	100.3±0.057	0.328	0.3
	10.0	10.00	0.0	100.00					
	14.0	13.99	0.07	99.93					
	20.0	19.99	0.05	99.95					
<b>XPM</b>	18.0	18.04	0.22	100.22	0.152	100.02±0.153	100.6±0.53	0.685	0.082
	28.0	27.98	0.11	99.89					
	35.0	35.03	0.08	100.08					
	45.0	44.96	0.08	99.92					

#### 4. CONCLUSION

A simple, selective, rapid and sensitive method has been proposed for the assay of drugs and in pharmaceutical formulations. The method is focused on the well-characterized and proven red-ox reaction and uses very simple, cheaper chemicals and easily accessible instruments. Other features like short performance time, ease of handling and the non-use of organic solvents also suggest this procedure as a routine laboratory method. The method is successfully applied to quantifying drugs in tablets and injecting them without intervention from common excipients. The current method is ideal for evaluation of BSZ, DVL, FMT, NFN and XPM in bulk drugs and pharmaceuticals. therefore, this method can be used in laboratories for quality control.

#### 5. ACKNOWLEDGMENT

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#### 6. REFERENCES

1. Naveen Kumar GS, Dinesh M, Gokul Nanda G and Hanumanthachar Joshi. Development and Validation of UV Spectrophotometric Method for Quantitative Estimation of Balsalazide From Capsule Formulation. World Journal of Pharmaceutical Research, 2019; 8(3): 1185-1190.
2. Naveen Kumar GS, Manohara YN and Channabasavaraj KP. Development and Validation of Spectrophotometric Methods for The Estimation of Balsalazide in Pharmaceutical Dosage Forms. International Journal of Chemical Science, 2008; 6(2): 497-502.
3. Salil K Bhattacharya, Parantapa Sen and Arunabha Ray. Pharmacology, 2003; 2: 334.
4. Prakash A and Spencer. Review of Pharmacology and Clinical Efficacy in Ulcerative

- Colitis. *Drugs*, 1998; 1998: 56 & 83.
5. Malathi S and Shivbalan S. Crohn's Disease, *Gastroenterology & Hepatology – I*, 2006; 73(8): 723-729.
  6. Ananda kumar K, Varadharajan K, Ayyappan T, Nageswara Rao P and Sujatha K. Estimation of Balsalazide in Bulk and in Formulation by UV-Visible Spectrophotometry. *Research Journal of Pharmacy and Technology*, 2008; 1(4): 472- 474.
  7. Eswara Rao Bammidi, Vaikuntarao Lakinani, and Pavitra Paila. Development and Validation of Stability Indicating HPLC Method for Balsalazide in Bulk Drug Development. *International Journal of Modern Chemistry and Applied Science*, 2016; 3(1): 315-322.
  8. Anandakumar K, Jothieswari D, Subathrai R, Varatharajan K and Sivanseyal G. Development and Validation of RP–HPLC Method for the Estimation of Balsalazide in Pure and in Pharmaceutical Dosage forms. *Research Journal of Pharmaceutical and Technology*, 2010; 3(1): 136-140.
  9. Kaja, RK, Surendranath KV, Radhakrishnanand P, Satish J, and Satyanarayana PVV. A Stability-Indicating LC Method for Analysis of Balsalazide Disodium in the Bulk Drug and in Pharmaceutical Dosage Forms. *Chromatographia*, 2009; 69(9-10): 1007–1012.
  10. Liebowitz MR, Yeung PP, Entsuah R. A randomized, double-blind, placebo-controlled trial of desvenlafaxine succinate in adult outpatients with major depressive disorder. *Journal of Clinical Psychiatry*, 2007; 68(11): 1663-72.
  11. Akram M, El-Didamony, Mona O, Abo-Elsoad. Utilization of Ion-Pair Complex Formation for The Spectrophotometric Determination of Some Antidepressant Drugs in Pharmaceutical Formulations. *International Journal of Pharmacy and Pharmaceutical Sciences*, 2016; 8(6): 188-194.
  12. Sreeja U, Gurupadayya BM, Chandan RS. Novel Spectrophotometric Methods for The Quantification of Desvenlafaxine in Pure and Pharmaceutical Dosage Form. *Asian Journal of Pharmaceutical and Clinical Research*, 2015; 8(2): 267-270.
  13. Ganesan, Abirami LR and Vetrichevan T. Spectrophotometric method for estimation of desvenlafaxine succinate in tablet dosage form. *Research Journal of Pharmaceutical, Biological and Chemical Sciences*, 2011; 2: 721-729.
  14. Dimal A Shah, Riddhi S Nakrani, Sunil LB, Usmangani KC and Kashyap KB. Development and Validation of HPTLC Method for the Estimation of Desvenlafaxine in Tablet Formulation. *Journal of Planar Chromatography*, 2012; 25: 174–177.
  15. Shubhangi MP and Sunil RD. Application of Stability Indicating High Performance Thin

- Layer Chromatographic Method for Quantitation of Desvenlafaxine in Pharmaceutical Dosage Form. *Journal of Liquid Chromatography & Related Technologies*, 2012; 35: 499–510.
16. Shah D, Nakrani R, Baldania S, Chhalotiya U and Bhatt K. Stability indicating liquid chromatographic method for the estimation of desvenlafaxine in pharmaceutical dosage form. *Chemical Industry and Chemical Engineering Quarterly*, 2011; 17(3): 341–348.
17. Qin F, Li N, Qin T, Zhang Y and Li F. Simultaneous quantification of venlafaxine and O-desmethylenlafaxine in human plasma by ultra-performance liquid chromatography–tandem mass spectrometry and its application in a pharmacokinetic study. *Journal of Chromatography B*, 2010; 878(7-8): 689–694.
18. Shubhangi MP, Laxman DK, Satish YG, Sunil RD. LC–UV and LC–MS evaluation of stress degradation behaviour of desvenlafaxine. *Journal of Pharmaceutical Analysis*, 2012; 2(4): 264–271.
19. Amit Anand, Dubey SK. New, sensitive and validated spectrofluorimetric method for the estimation of desvenlafaxine succinate in bulk and in pharmaceutical formulations. *World Journal of Pharmaceutical Sciences*, 2014; 2(2): 196-202.
20. Sanghavi, Bankim Srivastava, Ashwini. Adsorptive stripping differential pulse voltammetric determination of Venlafaxine and Desvenlafaxine employing nafion - carbon nanotube composite glassy carbon electrode. *Electrochimica Acta*, 2011; 56: 4188-4196.
21. Szafranski W, Cukier, Ramirez A, Menga G, Sansores R, Nahabedian S, Peterson S and olsson H. Efficacy and Safety of Budesonide/Formoterol in the Management of Chronic obstructive Pulmonary Disease. *European Respiratory Journal*, 2003; 21: 74- 81.
22. Duygu Taskin, Gizem E, Sıdıka SunGur. A validated spectrophotometric method for determination of formoterol fumarate dihydrate in bulk and dosage form using methyl orange as ion pair reagent. *Marmara Pharmaceutical Journal*, 2016; 20: 275-279.
23. Prasad AVSS. Simultaneous spectrophotometric determination of formoterol fumarate and budesonide in their combined dosage form. *Indian Journal of Chemical Technology*, 2006; 13: 81-83.
24. Assi KH, Tarsin W and Chrystyn H. High performance liquid chromatography assay method for simultaneous quantitation of formoterol and budesonide in Symbicort Turbuhaler. *Journal of Pharmaceutical and Biomedical Analysis*, 2006; 41(1): 325– 328.
25. Trivedi RK, Dhairyshil SC, Mukesh CP. A Rapid, Stability-Indicating RP-HPLC Method for the Simultaneous Determination of Formoterol Fumarate, Tiotropium Bromide, and

- Ciclesonide in a Pulmonary Drug Product. *Scientia Pharmaceutica*, 2012; 80(3): 591–603.
26. Samuel OA, Muhammad Asif. Validation of a RP-HPLC method for the assay of formoterol and its related substances in formoterol fumarate dihydrate drug substance. *Journal of Pharmaceutical and Biomedical Analysis*, 2003; 33: 935-945.
27. El-Bagary RI, Fouad MA, El-Shal MA and Tolba EH. Forced degradation of mometasone furoate and development of two RP-HPLC methods for its determination with formoterol fumarate or salicylic acid. *Arabian Journal of Chemistry*, 2016; 9(3): 493–505.
28. Parmar VK, Patel HN and Patel BK. Sensitive and Robust Methods for Simultaneous Determination of Beclomethasone Dipropionate and Formoterol Fumarate Dihydrate in Rotacaps. *Journal of Chromatographic Science*, 2014; 52(10): 1255–1266.
29. Jing-Zheng Song, Jue Chen, Song-Jiu Tian, Zeng-Pei Sun. Assay for the determination of low dosage form of formoterol dry syrup by capillary electrophoresis with head- column field-amplified sample stacking. *Journal of Pharmaceutical and Biomedical Analysis*, 1999; 21: 569–576.
30. Akapo SO, Wegner M, Mamangun A, McCrea C, Asif M and Dussex JC. Optimization and validation of a gas chromatographic method for analysis of (RS, SR)-diastereoisomeric impurity in formoterol fumarate. *Journal of Chromatography A*, 2004; 1045(1-2): 211–216.
31. Apperley DC, Harris RK, Larsson T and Malmstrom T. Quantitative nuclear magnetic resonance analysis of solid formoterol fumarate and its dihydrate. *Journal of Pharmaceutical Sciences*, 2003; 92(12): 2487–2494.
32. Jarvis B and Faulds D. Nelfinavir. *Drugs*, 1998; 56(1): 147–167.
33. Vanitha K, Prakash And Jangala Venkateswara Rao. Spectrophotometric Determination of Nelfinavir Mesylate. *E-Journal of Chemistry*, 2006; 3(2): 78-82.
34. Qiufang J, Yongjia Shen, Yanhui Tang, Fuzheng Ren, Xinhong Y and Zhian Hou. Determination of nelfinavir mesylate as bulk drug and in pharmaceutical dosage form by stability indicating HPLC. *Journal of Pharmaceutical and Biomedical Analysis*, 2006; 41: 1065–1069.
35. Van Heeswijk RP, Hoetelmans RM, Harms R, Meenhorst P, Mulder J, Lange JM and Beijnen J. Simultaneous quantitative determination of the HIV protease inhibitors amprenavir, indinavir, nelfinavir, ritonavir and saquinavir in human plasma by ion-pair high-performance liquid chromatography with ultraviolet detection. *Journal of Chromatography B- Biomedical Sciences and Applications*, 1998; 719(1-2): 159– 168.

36. Harumi Yamada, Hajime Kotaki, Tetsuya Nakamura, Aikichi Iwamoto. Simultaneous determination of the HIV protease inhibitors indinavir, amprenavir, saquinavir, ritonavir and nelfinavir in human plasma by high-performance liquid chromatography. *Journal of Chromatography B*, 2001; 755: 85–89.
37. Vanitha Prakash K, Venkateswararao J and Appala Raju N. RP-HPLC Method for the Estimation of Nelfinavir Mesylate in Tablet Dosage Form. *E-Journal of Chemistry*, 2007; 4(3): 302-306.
38. Neeraj Kaul, Himani Agrawal, Paradkar AR, Mahadik KR. Stability indicating high performance thin-layer chromatographic determination of nelfinavir mesylate as bulk drug and in pharmaceutical dosage form. *Analytica Chimica Acta*, 2004; 502: 31–38.
39. Tiwari RN and Bonde CG. LC, LC–MS/TOF and MSn studies for the identification and characterization of degradation products of nelfinavir mesylate. *Journal of Pharmaceutical and Biomedical Analysis*, 2011; 55(3): 435–445.
40. Jingduan Chi, Anura LJ, Judith AS, Toshiro Motoya and Francesca TA. Simultaneous determination of five HIV protease inhibitors nelfinavir, indinavir, ritonavir, saquinavir and amprenavir in human plasma by LC/MS/MS. *Journal of Pharmaceutical and Biomedical Analysis*, 2002; 30: 675–684.
41. Justesen US, Pedersen C, Klitgaard NA. Simultaneous quantitative determination of the HIV protease inhibitors indinavir, amprenavir, ritonavir, lopinavir, saquinavir, nelfinavir and the nelfinavir active metabolite M8 in plasma by liquid chromatography. *Journal of Chromatography B*, 2003; 783: 491–500.
42. Pravin BC, Jadhav AS, Ingale SK, Chemate SZ, Raskar MA. Development and validation of reverse phase liquid chromatographic methods for the determination of Nelfinavir mesylate in tablet form. *Der Pharmacia Lettre*, 2012; 4 (1): 135-141.
43. Sweetman S Ed. *Martindale, The Complete Drug Reference*; Pharmaceutical Press Mondin, 2009; 36th edition.
44. Mohamed Gaber, Abdalla MK, Ahmed S El-Kady. New and sensitive spectrophotometric method for determination of xipamide in pure and dosage forms by complexation with Fe(III), Cu(II), La(III), UO<sub>2</sub>(II), Th(IV) and ZrO(II) ions. *International Research of Pharmacy and Pharmacology*, 2011; 1(9): 215-220.
45. Wagieh, NE, Abbas SS, Abdelkawy M and Abdelrahman MM. Spectrophotometric and spectrodensitometric determination of triamterene and xipamide in pure form and in pharmaceutical formulation. *Drug Testing and Analysis*, 2010; 2: 113-121.
46. Omar MA. Spectrophotometric and Spectrofluorimetric Determination of Certain

- Diuretics Through Ternary Complex Formation with Eosin and Lead (II). *Journal of Fluorescence*, 2009; 20(1): 275–281.
47. Michael EE, Ahmad AA, Hesham Salem and Mahmoud AO. Spectrophotometric and spectrofluorimetric determination of certain diuretics in pure forms and in their pharmaceutical formulations. *Bull. Pharm. Sci*, 2006; 29(1): 33-58.
48. Walash MI, El-Enany N, Eid MI and Fathy ME. Stability–Indicating Spectrofluorimetric Methods for the Determination of Metolazone and Xipamide in Their Tablets, Application to Content Uniformity Testing. *Journal of Fluorescence*, 2013; 24(2): 363–376.
49. Abd El-Hay SS, Hashem H and Gouda AA. High performance liquid chromatography for simultaneous determination of xipamide, triamterene and hydrochlorothiazide in bulk drug samples and dosage forms. *Acta Pharmaceutica*, 2016; 66(1): 109–118.
50. Legorburu MJ, Alonso RM and Jimenez RM. Determination of The Non-Thiazide Diuretic Xipamide In Pharmaceuticals and Urine by HPLC With Amperometric Detection. *Journal of Liquid Chromatography & Related Technologies*, 1999; 22(5): 735–746.
51. Sane RT, Sadana GS, Bhounsule GJ, Gaonkar MV, Nadkarni AD and Nayak VG. High-performance liquid chromatographic determination of xipamide and clopamide in pharmaceuticals. *Journal of Chromatography A*, 1986; 356: 468–472.
52. Maher HM, Youssef RM, El-Kimary EI, Hassan EM and Barary MA. Bioavailability study of triamterene and xipamide using urinary pharmacokinetic data following single oral dose of each drug or their combination. *Journal of Pharmaceutical and Biomedical Analysis*, 2012; 61: 78–85.
53. Legorburu MJ, Alonso RM and Jimenez RM. Voltammetric study of the diuretic xipamide. *Bioelectrochemistry and Bioenergetics*, 1993; 32(1): 57–66.