

PYRIMIDINE HYDRAZONES AS POTENT ANTICONVULSANT

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Corresponding Author*Srivastav Sanchit**IEC Group of Institutions,
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of pharmacy.**ABSTRACT**

Pyrimidine constitutes an important class of heterocycles in drug discovery & is very well known for their anticancer, antimicrobial, antioxidant & antiviral activities. **Hydrazones** is a class of organic compounds with the structure $R_1R_2C=NNH_2$. They are formed usually by the action of hydrazine on ketones or aldehydes & have efficient CNS depressant, analgesics activity. In the same context, **Schiff bases** of pyrimidine hydrazones were prepared by the reaction of “**4-(4-Chloro-phenyl)-6-methyl-2-oxo-1,2,3,4-tetrahydro-pyrimidine-5-carboxylic acid hydrazide**” with **citral, camphor, furfuraldehyde**

& **menthone** individually by refluxing them in the presence of glacial acetic acid (as solvent) for at least 2hrs to yield their respective derivatives. Some of the compounds shown potent and efficient anticonvulsant action as mentioned in paper. All the drugs were tested on mice at 30mg/kg body weight and reported in the paper. In addition to this, the present studies revealed that the Schiff bases derivatives of Pyrimidine hydrazone could be used to synthesize the compounds having potent biological activities such as anticonvulsants. All the synthesized compounds were evaluated at the dose of 30mg/kg body weight using Isoniazid induced model and Thiosemicarbazide model for their anticonvulsant activity which were found to be pharmacologically active.

KEYWORD: Pyrimidine Hydrazones, Anticonvulsant pyrimidines, Pyrimidine Hydrazones as anticonvulsant etc.

1. INTRODUCTION

1.1 Pyrimidine: Heterocyclic aromatic organic compound like benzene or pyridine, having two nitrogen atoms at positions 1 and 3 of the six-membered ring; they have isomers in the forms of diazine.^[1]

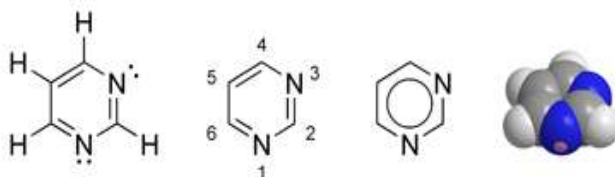


Figure 1

IUPAC Name: 1, 3-Diazine, m-Diazine

Formula: C₄H₄N₂

Molecular Mass: 80.088 g mol⁻¹

Density: 1.016 g cm⁻³

Solubility: Alcohol, Water

Melting Point: 20-22 °C

1.2 Types of Pyrimidine: Three nucleobases found in nucleic acids, cytosine (C), thymine (T), and uracil (U), are pyrimidine derivatives.^[1]

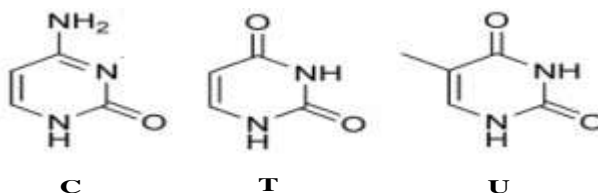


Figure 2

1.2 Hydrazone: An organic compounds having structure of R₁R₂C=NNH₂ and are associated to ketones and aldehydes by substitution of the oxygen by means of NNH₂ functional group. They are designed basically by the feat of hydrazine on ketones or aldehydes.^[2]

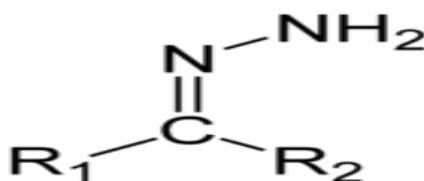
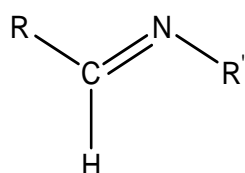


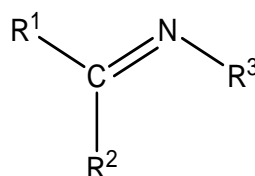
Figure 3

Hydrazones have antioxidant, antimicrobial, antimalarial, antiviral actions and if they are allowed to fuse with Pyrimidine they produce CNS activity too. Therefore the above data clearly showed that pyrimidine hydrazones are potent biologically active compounds.^[3]

1.3 Schiff base: A **Schiff base**, invented by Hugo Schiff, is a compound with a functional group that consists of a C=N double bond by means of nitrogen atom connected to an aryl or alkyl group and having general formula of $R^1R^2C=NR^3$, where R is an organic side chain. In this definition, *Schiff base* is identical to **azomethine**.^[4]



General structure of a Schiff base



General structure of an azomethine

Figure 4

2. MATERIALS AND METHODS

2.1 Synthesis of Pyrimidine hydrazone derivatives

Procedure: Synthesis of Pyrimidine hydrazone derivatives can be divided into following 3 steps.

Step (1): *Synthesis of ethyl 6-methyl-2-oxo-4-aryl-1,2,3,4-tetrahydro pyrimidin-5-carboxylates.*

Urea, ethylacetoacetate and aromatic aldehyde (0.1 mol) were mixed (equimolar amount) in ethanol (25 mL app.). A catalytic amount of conc. HCl (1ml) was added to the mixture, which was then refluxed for three hours. The contents were kept in refrigerator overnight. The solid separated out was filtered off. The filtrate was further refluxed on a water bath for 1.5 hour. A solid separated out on cooling was filtered and recrystallized from ethanol.

Where aromatic aldehyde used are as

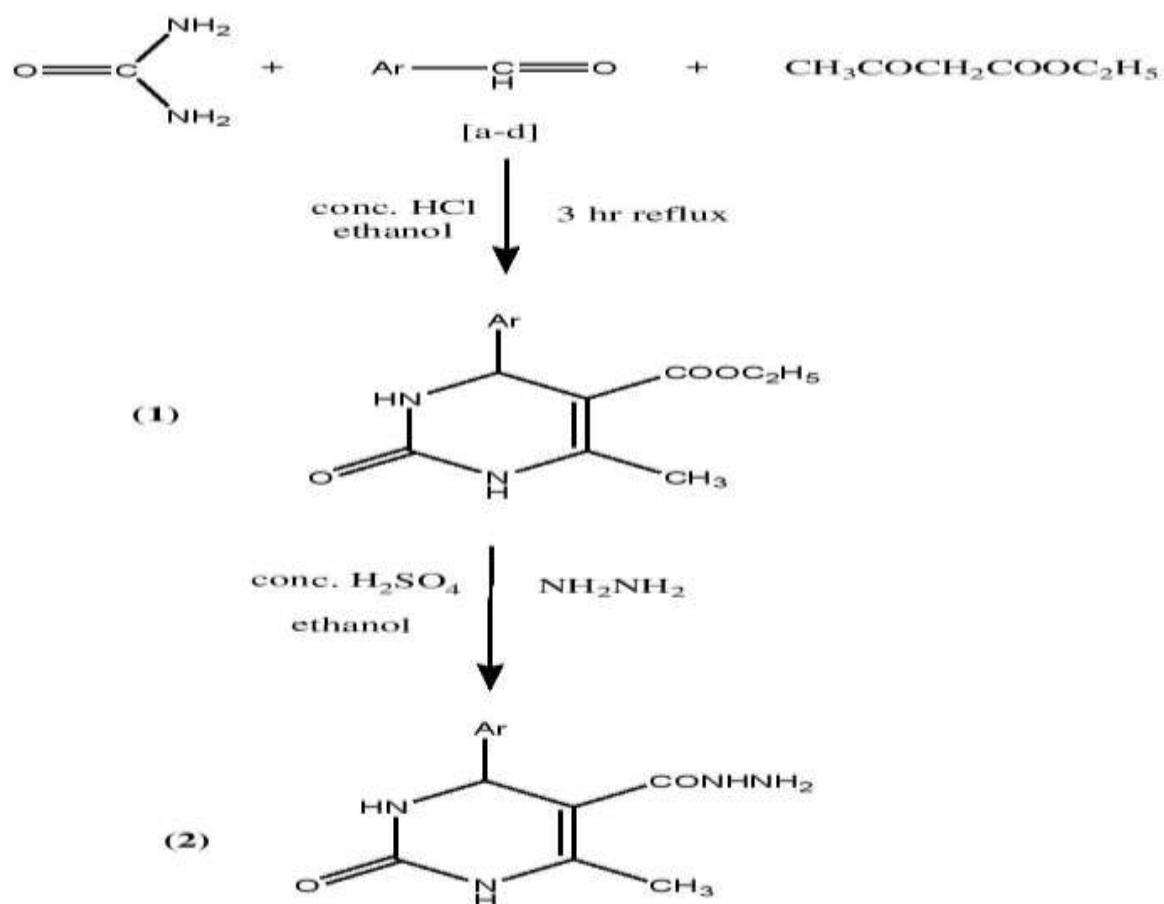
- 1). Benzaldehyde (C_6H_4CHO)
- 2). 6-Chloro-benzaldehyde (2-Cl- C_6H_4CHO)
- 3). p-Chloro-benzaldehyde (4-Cl- C_6H_4CHO)
- 4). p-Br-benzaldehyde (4-Br- C_6H_4CHO)

Step (2): Synthesis of 6-methyl-2-oxo-1,2,3,4-tetrahydropyrimidin-5-carbohydrazides

To 0.1 mol of (i) in ethanol (25 mL), hydrazine hydrate was added, followed by a catalytic amount of conc. H_2SO_4 (3 drops). The mixture was refluxed for two hours. Excess solvent was removed and, on cooling, a solid was formed. The solid was crystallized from ethanol.

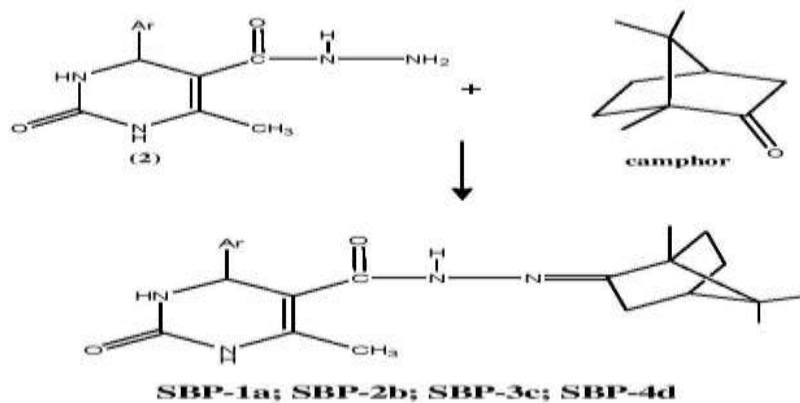
Step (3): Synthesis of various Schiff bases of the above obtained pyrimidine hydrazones

On adding the final product of above *step 2* with *Terpenes* like camphor, citral, menthone and furfuraldehyde, we obtain following desired Schiff bases for the derivatives of Pyrimidine Hydrazone (*i.e.* **A**, **B**, **C** & **D**) respectively as per proposed scheme which are described & elaborated below on the synthetic scheme part

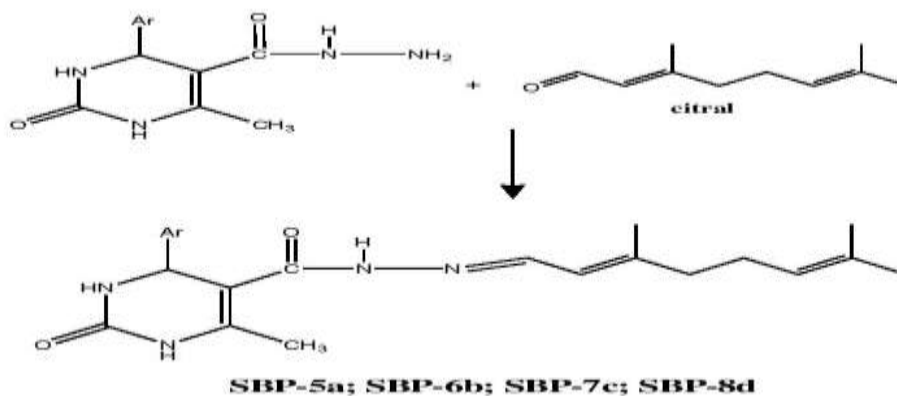


Scheme 1: Series of desired pyrimidine hydrazone derivatives will be synthesized as follows **Where:** **Ar** = **a:** C_6H_5 ; **b:** $4\text{-Br-C}_6\text{H}_5$; **c:** $2\text{-Cl-C}_6\text{H}_4$; **d:** $4\text{-Cl-C}_6\text{H}_3$

A).



B).



Step 3: On adding the final product of above *step ii* with camphor & citral to yield the desired Schiff bases for the derivatives of Pyrimidine Hydrazone respectively as per scheme which are described & elaborated below on the synthetic scheme part

Where: Ar = a: C₆H₅; b: 4-Br-C₆H₄; c: 2-Cl-C₆H₄; d: 4-Cl-C₆H₄;

3. PHYSICOCHEMICAL DATA

Table 1

Compound Name	Ar	Ketone Type	M.P. (°C)	% Yield	Molecular Formula	Molecular Weight	R _f	Color
SBP-1a	C ₆ H ₅	Camphor	195	65%	C ₂₂ H ₂₈ N ₄ O ₂	380.48	0.75	White
SBP-2b	4-Br-C ₆ H ₄	Camphor	210	73%	C ₂₂ H ₂₇ BrN ₄ O ₂	459.38	0.72	Yellow
SBP-3c	2-Cl-C ₆ H ₄	Camphor	195	61%	C ₂₂ H ₂₇ ClN ₄ O ₂	414.93	0.63	Yellow
SBP-4d	4-Cl-C ₆ H ₄	Camphor	185	65%	C ₂₂ H ₂₇ ClN ₄ O ₂	414.93	0.68	White
SBP-5a	C ₆ H ₅	Citral	65	92%	C ₂₂ H ₂₈ N ₄ O ₂	380.48	0.63	Black
SBP-6b	4-Br-C ₆ H ₄	Citral	205	88%	C ₂₂ H ₂₇ BrN ₄ O ₂	458.13	0.81	Yellow
SBP-7c	2-Cl-C ₆ H ₄	Citral	63	76%	C ₂₂ H ₂₇ ClN ₄ O ₂	414.93	0.82	Black

4. PHARMACOLOGICAL EVALUATION

4.1 Anticonvulsant activity

4.1.1 Requirements: All the synthesized compounds were evaluated for their anticonvulsant activity using Isoniazid; Thiosemicarbazide induced convulsion models on male albino mice

(20–25g). PEG-400 & DMSO is used as vehicle for Thiosemicarbazide induced model and INH induced model respectively, whereas Diazepam as a standard drug.

4.1.2 Chemical induced models: 10 mice of either sex with a weight of 22-25g were treated with the test compounds or the standard (e.g. diazepam 10 mg/kg i.p.) by intraperitoneal administration. Controls received the vehicle only. 30 min after i.p. treatment the animals were injected with a subcutaneous dose of 300 mg/kg Isoniazid (isonicotinic acid hydrazide & thiosemicarbazide (20 mg/kg). The occurrence of clonic seizures, tonic seizures and death or recovery was recorded after 0.5hr, 1hr, and 2hr respectively ^[5-8]. Not Protected means death of the rats occurs at the mentioned time.

4.1.3 Evaluation of Anticonvulsant activity: INH induced convulsion model

Table 2

Treatment	.5hr	1hr	2hr
Control	_____	_____	_____
SBP-1a	30mg	30mg	<i>Not Protected</i>
SBP-2b	30mg	<i>Not Protected</i>	<i>Not Protected</i>
SBP-3c	30mg	30mg	<i>Not Protected</i>
SBP-4d	30mg	<i>Not Protected</i>	<i>Not Protected</i>
SBP-5a	<i>Not Protected</i>	<i>Not Protected</i>	<i>Not Protected</i>
SBP-6b	30mg	<i>Not Protected</i>	<i>Not Protected</i>
SBP-7c	30mg	<i>Not Protected</i>	<i>Not Protected</i>
SBP-8d	30mg	30mg	<i>Not Protected</i>
Diazepam	10mg	10mg	10mg

4.1.4 Evaluation of Anticonvulsant activity: Thiosemicarbazide Induced convulsion model

Table 3

Treatment	.5hr	1hr	2hr
Control	_____	_____	_____
SBP-1a	30mg	30mg	<i>Not Protected</i>
SBP-2b	30mg	<i>Not Protected</i>	<i>Not Protected</i>
SBP-3c	30mg	30mg	<i>Not Protected</i>
SBP-4d	30mg	<i>Not Protected</i>	<i>Not Protected</i>
SBP-5a	<i>Not Protected</i>	<i>Not Protected</i>	<i>Not Protected</i>
SBP-6b	30mg	<i>Not Protected</i>	<i>Not Protected</i>
SBP-7c	30mg	<i>Not Protected</i>	<i>Not Protected</i>
SBP-8d	30mg	<i>Not Protected</i>	<i>Not Protected</i>
Diazepam	10mg	10mg	10mg

5. RESULTS AND DISCUSSION

All the synthesized compounds were evaluated for their anticonvulsant activity using various chemical induced convulsion models on male albino mice (20–25g) and diazepam 10mg/kg body weight as a standard drug. Table no 2 and 3 All the synthesized derivatives were evaluated at the dose of 30mg/kg body weight & have shown good anticonvulsant activity & the compounds *SBP-1a* and *SBP-3c* was found to be most active amongst all the screened compounds using both the models i.e. Isoniazid induced model as well as Thiosemicarbazide model.

6. NEUROTOXICITY SCREENING

6.1 Rota rod test: Minimal motor impairment was measured in mice by the Rotarod test. The mice were trained to stay on an accelerating Rotarod that rotates at 6 revolutions per minute. The rod diameter was 3.2 cm. neurotoxicity was indicated by the inability of the animal to maintain equilibrium on the rod for at least 1 min in each of three trials. The dose at which the animals were unable to grasp the Rotarod was determined.

6.1.1 Evaluation for Neurotoxicity by Rota-Rod model

Table 4

Treatment	Neurotoxicity(yes/no) at 20 mg
Control	_____
SBP-1	Non-Neurotoxic
SBP-2	Non-Neurotoxic
SBP-3	Non-Neurotoxic
SBP-4	Non-Neurotoxic
SBP-5	Non-Neurotoxic
SBP-6	Non-Neurotoxic
SBP-7	Non-Neurotoxic
SBP-8	Non-Neurotoxic

7. RESULT AND DISCUSSIONS

Activity of the drugs interfering with motor coordination was checked by the Rota rod for neurotoxicity test. None of the compounds was found to be neurotoxic at a dose of 20 mg/kg body weight amongst all the tested compounds.

8. CONCLUSION

All the synthesized compounds were evaluated at the dose of 30mg/kg body weight using Isoniazid induced model and Thiosemicarbazide model for their anticonvulsant activity using various chemical induced convulsion models on male albino mice (20–25g) and diazepam

10mg/kg body weight as a standard drug. The compounds **SBP-1a** and **SBP-3c** were found to be most active amongst all the screened compounds.

9. ACKNOWLEDGMENT

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